

# New Acylphloroglucinol Derivatives with Diverse Architectures from *Hypericum henryi*

Xing-Wei Yang,<sup>†,§,||</sup> Yuanqing Ding,<sup>‡,||</sup> Jing-Jing Zhang,<sup>†,§,||</sup> Xia Liu,<sup>†,§</sup> Li-Xin Yang,<sup>†</sup> Xiao-Nian Li,<sup>†</sup> Daneel Ferreira,<sup>‡</sup> Larry A. Walker,<sup>‡</sup> and Gang Xu<sup>\*,†</sup>

<sup>†</sup>State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, Chinese Academy of Sciences, Kunming 650201, Yunnan P. R. China

<sup>‡</sup>National Center for Natural Products Research, Research Institute of Pharmaceutical Science, Department of Pharmacognosy, and Department of Pharmacology, School of Pharmacy, University of Mississippi, University, Mississippi 38677, United States

<sup>§</sup>University of Chinese Academy of Sciences, Beijing 100049, P. R. China

Supporting Information

**ABSTRACT:** Hyphenrones A–F (1–6), six polycyclic polyprenylated acylphloroglucinol derivatives with four architectures including three unprecedented cores as exemplified by 1, 3, and 4, were isolated from *Hypericum henryi*. Compounds 3 and 4 possess two unique 5/8/5 and 6/6/5/8/5 fused ring systems, respectively. Their absolute configurations were defined by experimental and calculated ECD of 4 and X-ray diffractions of 5 and 6, coupled with their putative biosynthetic origins. Three compounds exhibited interesting AChE inhibitory activities.

**P** lants of the genus *Hypericum*, widely used as traditional medicine in many countries,<sup>1</sup> are a rich source of polycyclic polyprenylated acylphloroglucinols (PPAPs) with diverse structures and significant bioactivities.<sup>2</sup> PPAPs are a group of structurally fascinating and synthetically challenging natural products possessing highly oxygenated acylphloroglucinol-derived cores densely decorated with prenyl substituents, which collectively exhibit a remarkably broad range of biological activities such as tumor inhibitive, antimicrobial, HIV preventative, antioxidant, and antidepressant.<sup>2,3</sup> Since the characterization of hyperforin, a typical PPAP type molecule with cognitive-enhancing and memory-facilitating properties as well as neuroprotective effects against Alzheimer's disease,<sup>4</sup> approximately 200 PPAPs have been reported, most of which are *endo*-bicyclic polyprenylated acylphloroglucinols (*endo*-BPAPs) with a bicyclo[3.3.1]nonane-2,4,9-trione core.<sup>2</sup>

Hypericum henryi is a traditional Chinese medicinal plant used for the treatment of hepatitis in Yunnan, China.<sup>5</sup> As a part of our systematic search for new and bioactive natural PPAPs from the Hypericum genus,<sup>6</sup> hyphenrones A–F (1–6), six PPAP-type derivatives with four architectures including the three unprecedented cores as exemplified by 1, 3, and 4, together with three known precursors 7–9, were isolated from H. henryi (Figure 1 and Scheme 1). Their absolute configurations were defined by experimental and calculated electronic circular dichroism (ECD) spectra of 4 and the X-ray diffractions (Cu K $\alpha$ ) data of 5 and 6 coupled with the presumed close relationship in their biosynthetic pathway. Compounds 1 and 2 possess an unprecedented *seco*-PPAP skeleton formed by cleavage of the C-1/C-9 bond of the normal *endo*-BPAPs. Compounds 3 and 4, featuring fascinating





Figure 1. Structures of compounds 1-4.

5/8/5 and 6/6/5/8/5 fused ring systems, were presumably synthetically derived from 1 and 2 via key aldol condensation and Diels–Alder cycloaddition reactions, respectively. Compounds 1, 3, and 4 exhibited interesting AChE inhibitory activities. Herein, the structural elucidation including absolute configurational analysis and bioactive evaluation of 1-6 are reported.

Hyphenrone A (1) was obtained as a light yellow gum. Its molecular formula  $C_{35}H_{52}O_5$  was established by  $^{13}C$  NMR and HR-ESI-MS data (m/z 575.3704,  $[M + Na]^+$ ) indicating 10 indices of hydrogen deficiency. The  $^{13}C$  NMR and DEPT spectra revealed 35 carbon resonances corresponding to six quaternary carbons (including one oxygenated and three carbonyls), two methines, two methylenes, one methyl, and 24 other resonances assignable to an isobutyryl and four prenyl groups (Table S7, Supporting Information). Analysis of these data indicated the characteristic resonances of three non-conjugated carbonyls at  $\delta_C$  197.9 (C-2), 206.9 (C-4), and 206.6

Received:March 18, 2014Published:April 16, 2014

# Scheme 1. Putative Biosynthesis Pathways to $1-9^a$



<sup>a</sup>Key: red type, newly formed rings in unprecedented skeletons; blue rectangles, ORTEP drawings.

(C-10), two quaternary carbons at  $\delta_{\rm C}$  57.3 (C-5) and 48.7 (C-8), one methine at  $\delta_{\rm C}$  40.7 (C-7), and one methylene at  $\delta_{\rm C}$  37.1 (C-6) of a PPAP type metabolite.<sup>6</sup> Comparison of its 1D NMR data with those of related normal endo-BPAPs such as 3hydroxyhyperforin-3,9-hemiketal (9, Scheme 1), $^7$  a known PPAP that was also isolated in this study, revealed some significant differences. The resonances of the hemiketal carbon (C-9,  $\delta_{\rm C}$  108.1) and the characteristic quaternary carbon (C-1,  $\delta_{\rm C}$  71.4) in 9 were replaced by the resonances of an ester carbonyl ( $\delta_{\rm C}$  172.0) and a methine ( $\delta_{\rm C}$  63.2,  $\delta_{\rm H}$  4.54), respectively, in 1. Analysis of the 2D NMR data indicated that these two resonances could be attributed to C-9 and C-1, respectively, due to the HMBC correlations from the methine proton at  $\delta_{\rm H}$  4.54 (s) to C-2, C-3 ( $\delta_{\rm C}$  95.6), C-7, and C-8, from  $H_2$ -22 ( $\delta_H$  2.55 and 2.10) to C-4, C-5, C-6, and the ester carbonyl ( $\delta_{\rm C}$  172.0), and from Me-32 ( $\delta_{\rm H}$  1.09, s) to C-1 and C-8. An internal ester bridge between C-3 and C-5 was indicated by the indices of hydrogen deficiency along with the characteristic chemical shift of C-3 at  $\delta_{\rm C}$  95.6. On the basis of these results, compound 1 was presumably formed by cleavage of the C-1/C-9 bond of 9 followed by the formation of a fivemembered lactone moiety. Additional 2D NMR data including <sup>1</sup>H–<sup>1</sup>H COSY and HMBC confirmed the structure (Figure 2). Therefore, the molecular structure of 1 possesses an unprecedented seco-PPAP skeleton.

In the ROESY spectrum of 1, diagnostic cross-peaks observed for Me-32/H<sub>2</sub>-27, H<sub>2</sub>-33/H-1, and H-1/H-7 demon-



Figure 2. Key HMBC and <sup>1</sup>H-<sup>1</sup>H COSY correlations of 1, 3, and 4.

strated that Me-32 and CH<sub>2</sub>-27 were both  $\beta$ -oriented while H-1, H-7, and C-33 were all  $\alpha$ -oriented. The NOE correlations of H-1/H-17, Me-32/H-6 $\beta$ , and H-6 $\alpha$ /H-22 defined the  $\alpha$ orientations of CH<sub>2</sub>-22 and CH<sub>2</sub>-17. The same orientations of the prenyl groups at C-3 and C-5 were in agreement with the presence of a five-membered lactone moiety (Figure 3).



Figure 3. Key ROESY correlations of 1 and 4.

Analyses of the MS and NMR data (Table S8, Supporting Information) showed that hyphenrone B (2) shared the same carbon skeleton and relative configuration as 1. The isopropyl group in 1 was replaced by a phenyl group in 2.

The molecular formula of hyphenrone C (3) was determined to be  $C_{32}H_{44}O_5$  by its HR-ESI-MS (m/z 531.3080, [M + Na]<sup>+</sup>) and <sup>13</sup>C NMR data. Analysis of its 1D and 2D NMR data (Table S9, Supporting Information) revealed that 3 was similar to 1, especially the resonances of the eight-membered core and lactone moieties (rings A and B). However, comparison of their NMR data indicated that the C-10 carbonyl resonance ( $\delta_C$ 206.6) and the C-33 isoprenyl unit in 1 were absent in 3. Instead, an unusual  $\alpha,\beta$ -unsaturated formyl group ( $\delta_C$  190.0,  $\delta_H$ 10.12, s;  $\delta_C$  140.0, C- $\alpha$ , and  $\delta_C$  168.2, C- $\beta$ ) was observed in 3. This observation implied that 3 might be a trinor derivative of 1, which was most likely formed via oxidative cleavage of the C-35/C-36 double bond in 1 followed by aldol condensation between C-34 and C-10 to form the unusual cyclopentene Cring in 3. This assumption was confirmed by analysis of its 2D NMR data. The cyclopentene C-ring was elucidated on the basis of the HMBC correlations of Me-32 ( $\delta_{\rm H}$  0.79) with C-1 ( $\delta_{\rm C}$  66.5), C-8 ( $\delta_{\rm C}$  55.0), and C-33 ( $\delta_{\rm C}$  44.7), the formyl proton ( $\delta_{\rm H}$  10.12, H-35) with C-10, C-33, and C-34, and H-1 ( $\delta_{\rm H}$  4.32, d, J = 3.0 Hz) with the  $\alpha$  (C-34) and  $\beta$  (C-10) carbons of the  $\alpha,\beta$ -unsaturated formyl group. An isopropyl group at C-10 was indicated by the HMBC correlations from both Me-12 and Me-13 ( $\delta_{\rm H}$  1.30 and 1.23, respectively, d, J = 7.2 Hz) to C-10 as well as the proton spin system consisting of Me-12/H-11/Me-13 present in the <sup>1</sup>H-<sup>1</sup>H COSY spectrum (Figure 2). On the basis of similar correlations of C-1, C-3, C-5, C-7, and C-8 of 3 and 1 were identical. Thus, the structure of 3 possesses a new scaffold featuring a 5/8/5 fused ring system (Figure 2).

Hyphenrone D (4), a colorless gum, exhibited a molecular ion at m/z 584.3497 in its positive HR-EI-MS, which in conjunction with the <sup>13</sup>C NMR data, indicated a molecular formula of  $C_{38}H_{48}O_5$ , i.e., two hydrogens fewer than 2. The <sup>1</sup>H and <sup>13</sup>C NMR spectra of 4 (Table S10, Supporting Information) differed from those of 2 despite some similarities in the resonances of rings A and B along with their substituents. An ortho-disubstituted benzene ring in 4 was evident from resonances at  $\delta_{\rm H}$  7.37 (d, J = 7.8 Hz), 7.53 (t, J = 7.8 Hz), 7.38 (t, J = 7.8 Hz), and 7.87 (d, J = 7.8 Hz) compared to the monosubstituted benzene ring in 2. In addition, the resonances characteristic of C-1 ( $\delta_{\rm C}$  57.9, d;  $\delta_{\rm H}$  5.53, s) and the double bond ( $\delta_{\rm C}$  123.7, C-35, and 135.3, C-36) of the C-8 homoprenyl group in 2 were replaced by two quaternary carbons at  $\delta_{\rm C}$  78.4 and 37.4 and a methine carbon ( $\delta_{\rm C}$  56.6;  $\delta_{\rm H}$  3.34, dd, J = 11.0, 9.0 Hz) in 4. These observations suggested that 4 might be derived from 2 via an intramolecular cyclization, leading to the formation of two extra carbocycles, such cyclization occurring between C-1, C-35, C-36, and the carbons of the benzoyl group. This assumption was confirmed by the 2D NMR data. The HMBC correlations from Me-32 ( $\delta_{\rm H}$  1.29, s) to C-1, C-8  $(\delta_{\rm C}$  56.3), and C-33  $(\delta_{\rm C}$  40.5), from H-35  $(\delta_{\rm H}$  3.34) to C-1 and C-8, and from H-34 to C-1 coupled with the <sup>1</sup>H-<sup>1</sup>H COSY correlations of H-33/H-34/H-35 established the presence of the cyclopentene C-ring (Figure 2). The presence of the D-ring was confirmed by the HMBC correlations of H-35 with a benzoyl carbonyl carbon at  $\delta_{C}$  199.7 (C-10), of Me-37 and Me-38 ( $\delta_{\rm H}$  1.43 and 0.98) with C-35 ( $\delta_{\rm C}$  56.6), C-36 ( $\delta_{\rm C}$  37.4), and C-12 ( $\delta_{\rm C}$  151.5), of H-16 ( $\delta_{\rm H}$  7.87) with C-10, and of H-13 ( $\delta_{\rm H}$ 7.37) with C-36. Thus, the molecular structure of 4 possesses an unprecedented carbon skeleton comprising a 6/6/5/8/5 fused ring system.

The relative configurations of C-5, C-7, and C-8 of 4 were the same as in compounds 1–3, based on the ROESY correlations of Me-32/H-27, Me-32/H-6 $\beta$ , and H-6 $\alpha$ /H-22, that indicated  $\beta$ -orientations of Me-32 and CH<sub>2</sub>-27 and an  $\alpha$ orientation of CH<sub>2</sub>-22. In addition, the correlation between the olefinic protons H-18 ( $\delta_{\rm H}$  4.12) and H-23 ( $\delta_{\rm H}$  4.82) confirmed the  $\alpha$ -orientations of the isoprenyl groups at C-3 and C-5. The NOE correlations of Me-32/H-34 $\beta$  and H-34 $\alpha$ /H-35 indicated the  $\alpha$ -orientation of H-35 (Figure 3). However, the definition of the C-1 configuration was more problematic and had to be derived from its putative biosynthesis pathway. The  $\alpha$ orientation of the C-8 homoprenyl group in precursor **2** necessitated the  $\beta_i\beta_j$ -junction of ring D to the C-ring, which indicated the relative configuration of C-1 (Scheme 1).

The absolute configuration of 4 was confirmed by comparison of experimental and TD-DFT caculated ECD

spectra.<sup>8</sup> First, an arbitrarily assigned 1R,3S,5R,7S,8R,35R configuration was employed for the conformationally random search using the MMFF94s force field in the MacroModel software package. By using the Gaussian09 software package, 23 conformers, which possess the lowest energies from the conformational search with an energy cutoff of 17 kJ/mol (approximately 4 kcal/mol), were included for full geometry optimization at the B3LYP/6-31G\*\* level in the gas phase where 18 conformers were relocated. Further calculations were performed at the B3LYP/6-311++G\*\* and B3LYP-SCRF-(COSMO)/6-311++G\*\*//B3LYP/6-311++G\*\* levels in the gas phase and in MeOH, respectively. Conformational analysis and ECD calculations were performed at the same levels in the gas phase and in MeOH, respectively. The calculated individual and weighted ECD spectra of the predominant conformers were similar and match very well with the experimentally observed spectra at the above levels in both the gas phase and in MeOH (Figure 4). Therefore, the absolute configuration of 4 was defined to be 1R,3S,5R,7S,8R,35R.



**Figure 4.** Calculated and experimental ECD spectra of **4** (red and black, at the B3LYP/6-31G\*\* and B3LYP/6-311++G\*\* levels in the gas phase, respectively; olive, at the B3LYP-SCRF(COSMO)/6-311++G\*\*//B3LYP/6-311++G\*\* level in MeOH; blue, experimentally observed in MeOH).

Hyphenrone E (5) was determined to be a homologue of oxepahyperforin via analysis of the MS and NMR data (Table S11, Supporting Information).<sup>7</sup> The isopropyl moiety in oxepahyperforin was replaced by an isobutyl group in 5 (Scheme 1).

It is challenging to assign the absolute configuration of PPAPs because they are typically isolated as gums. However, crystals of 5 (CCDC 962624) and 6 (CCDC 962625) suitable for X-ray diffraction analysis were obtained from methanol. The final refinement on the Cu K $\alpha$  data of the crystal of 5 [the Flack parameter is 0.09(19) and the Hooft parameter is 0.11(5) for 2319 Bijvoet pairs] and 6 [the Hooft parameter is 0.11(7) for 1815 Bijvoet pairs] allowed an unambiguous assignment of the absolute structures as shown in Scheme 1,<sup>9</sup> indicating that the absolute configurations of C-1, C-3, C-5, C-7, and C-8 in 5 and 6 were consistent with the conclusions derived from the experimental and calculated ECD spectra of 4 and those of hyperforin established previously.<sup>2</sup> In addition, the biosyntheses of the members in Scheme 1 are presumably closely related (vide infra). Thus, the absolute configurations of these PPAPs should show a considerable degree of similarity.

Biosynthetically, PPAPs are presumably derived from a "mixed" prenylation/polyketide biosynthesis pathway (Scheme 1). Their acylphloroglucinol core structure may be derived from a characteristic polyketide-type biosynthesis involving the condensation of one acyl-CoA and three malonyl-CoA units.<sup>2,10</sup>

Prenylation of this core moiety would afford monocyclic polyprenylated acylphloroglucinols (MPAPs), which may be further cyclized to PPAP type metabolites with diverse carbon skeletons.<sup>2,10</sup> In this study, hyperform (7) and adhyperform (8)represented the normal endo-BPAPs with a bicyclo[3.3.1]nonane-2,4,9-trione core and should be the precursors of the 2,35-epoxy derivative hyphenrone E(5) and hemiketal derivative 9. Compounds 1 and 2 were likely formed by cleavage of the C-1/C-9 bond of hemiketal derivatives such as 9 via a retro-aldol mechanism and subsequent formation of a fivemembered lactone moiety. The intriguing structure of hyphenrone C (3) possessing a new scaffold featuring a 5/8/5 fused ring systems may be derived from 1 via oxidative cleavage of the C-35/C-36 double bond followed by formation of the cyclopentene ring via aldol condensation. The structure of hyphenrone D (4), featuring an unprecedented 6/6/5/8/5 fused ring system, was presumbly formed via an intermolecular Diels-Alder cycloaddition of 2. In addition, the known hyphenrone F (6), possessing a 9-nor-PPAP skeleton, may be derived from 1 via sequential hydrolysis and decarboxylation steps.<sup>11</sup> Therefore, the four architectures of these PPAPs are explicable in terms of their presumed biosynthetic origins, and their absolute configurations should accordingly be interrelated.

Several cytotoxic PPAPs have been previously isolated in our laboratory,<sup>6,12</sup> and these types of metabolites are associated with neurodegenerative diseases, such as Alzheimer's disease,<sup>4</sup> we examined the inhibitory activities of the aforementioned compounds against acetylcholinesterase (AChE) and the five human tumor cell lines HL-60, A-549, SMMC-7721, MCF-7, and SW480. Compounds 4 and 5 exhibited moderate cytotoxic activities (IC<sub>50</sub> 4.7–25.5  $\mu$ M, Table S6, Supporting Information) against the cancer cell lines using the MTT method.<sup>13</sup> In the AChE inhibition assay using the Ellman method,<sup>14</sup> compound **5** exhibited weak AChE inhibitory activity with an IC<sub>50</sub> value of 25.4  $\mu$ M. Notably, the PPAPs that possess a lactone moiety (1, 3, and 4) exhibited significant negative inhibitory activities (approximately –100% for each) at 50  $\mu$ M (Table S4, Supporting Information).

### ASSOCIATED CONTENT

#### **S** Supporting Information

Computational details of **4**, experimental procedures, physicalchemical properties, MS and NMR spectra for all new compounds, crystallographic data (CIF) for **5** and **6**. This material is available free of charge via the Internet at http:// pubs.acs.org.

## AUTHOR INFORMATION

#### **Corresponding Author**

\*E-mail: xugang008@mail.kib.ac.cn.

#### **Author Contributions**

<sup>||</sup>These authors contributed equally.

# Notes

The authors declare no competing financial interest.

#### ACKNOWLEDGMENTS

This work was financially supported by the foundations from NSFC (20972167), the Young Academic Leader Raising Foundation of Yunnan Province (No. 2009CI073), the foundation from CAS to G.X., and the US Department of Agriculture, Agricultural Research Service Specific Cooperative

Agreement (No. 58-6408-2-0009). We also thank the Mississippi Center for Supercomputing Research (MCSR) for the computational facilities.

## REFERENCES

(1) Avato, P. In *Studies in Natural Products Chemistry*; Rahman, A., Ed.; Elsevier: Netherland, 2005; Vol. 30, pp 603–634.

(2) Ciochina, R.; Grossman, R. B. Chem. Rev. 2006, 106, 3963-3986.
(3) (a) Singh, I. P.; Bharate, S. B. Nat. Prod. Rep. 2006, 23, 558-591.
(b) Singh, I. P.; Sidana, J.; Bharate, S. B.; Foley, W. J. Nat. Prod. Rep. 2010, 27, 393-416. (c) Biber, N.; Mows, K.; Plietker, B. Nat. Chem. 2011, 3, 938-942. (d) Zhang, Q.; Mitasev, B.; Qi, J.; Porco, J. A., Jr. J. Am. Chem. Soc. 2010, 132, 14212-14215. (e) Qi, J.; Beeler, A. B.; Zhang, Q.; Porco, J. A., Jr. J. Am. Chem. Soc. 2010, 132, 13642-13644.
(f) Richard, J.-A.; Pouwer, R. H.; Chen, D. Y.-K. Angew. Chem., Int. Ed. 2012, 51, 4536-4561. (g) Sparling, B. A.; Moebius, D. C.; Shair, M. D. J. Am. Chem. Soc. 2013, 135, 644-647. (h) George, J. H.; Hesse, M. D.; Baldwin, J. E.; Adlington, R. M. Org. Lett. 2010, 12, 3532-3535.
(4) Griffith, T. N.; Varela-Nallar, L.; Dinamarca, M. C.; Inestrosa, N. C. Curr. Med. Chem. 2010, 17, 391-406.

(5) Lan, M. In South Yunnan Materia Medica; Wu, Z. Y., Gao, L., Ed.; Science Press: Yunnan, Kunming, 2008; Vol. 1, pp 236–240.

(6) (a) Yang, X. W.; Deng, X.; Liu, X.; Wu, C. Y.; Li, X. N.; Wu, B.; Luo, H. R.; Li, Y.; Xu, H. X.; Zhao, Q. S.; Xu, G. *Chem. Commun.*2012, 48, 5998–6000. (b) Liu, X.; Yang, X. W.; Chen, C. Q.; Wu, C. Y.; Zhang, J. J.; Ma, J. Z.; Wang, H.; Yang, L. X.; Xu, G. J. Nat. Prod.
2013, 76, 1612–1618.

(7) Verotta, L.; Appendino, G.; Jakupovic, J.; Bombardelli, E. J. Nat. Prod. 2000, 63, 412–415.

(8) (a) Diedrich, C.; Grimme, S. J. Phys. Chem. A 2003, 107, 2524–2539. (b) Berova, N.; Di Beri, L.; Pescitelli, G. Chem. Soc. Rev. 2007, 36, 914–931. (c) Ding, Y.; Li, X. C.; Ferreira, D. J. Org. Chem. 2007, 72, 9010–9017.

(9) (a) Flack, H. D. Acta Crystallogr. **1983**, A39, 876–881. (b) Hooft, R. W. W.; Straver, L. H.; Spek, A. L. J. Appl. Crystallogr. **2008**, 4, 96–103.

(10) Adam, P.; Arigoni, D.; Bacher, A.; Eisenreich, W. J. Med. Chem. 2002, 45, 4786–4793.

(11) Verotta, L.; Lovaglio, E.; Sterner, O.; Appendino, G.; Bombardelli, E. *Eur. J. Org. Chem.* **2004**, 1193–1197.

(12) (a) Xu, G.; Feng, C.; Zhou, Y.; Han, Q. B.; Qiao, C. F.; Huang, S. X.; Chang, D. C.; Zhao, Q. S.; Luo, K. Q.; Xu, H. X. *J. Agric. Food Chem.* **2008**, *56*, 11144–11150. (b) Xu, G.; Kan, W. L. T.; Zhou, Y.; Song, J. Z.; Han, Q. B.; Qiao, C. F.; Cho, C. H.; Rudd, J. A.; Lin, G.; Xu, H. X. *J. Nat. Prod.* **2010**, *73*, 104–108.

(13) Alley, M. C.; Scudiero, D. A.; Monks, A.; Hursey, M. L.; Czerwinski, M. J.; Fine, D. L.; Abbott, B. J.; Mayo, J. G.; Shoemaker, R. H.; Boyd, M. R. *Cancer Res.* **1988**, *48*, 589–601.

(14) Ellman, G. L.; Courtney, K. D.; Andres, V., Jr.; Featherstone, R. M. Biochem. Pharmacol. **1961**, *7*, 88–95.